

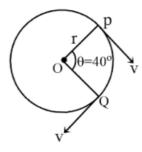
Max	. Marks	: 60						Date: 27.11.20)22		
				JB 2 M PHYSICS : Topics: Ci	,	ET A)					
1.	A pla	ne is revolving a	round the e	earth with a speed	l of 100 km/l	n at a constant hei	ght from th	e surface of the ear	rth		
	The c	change in the velo	city as it tr	avels half-circle is	S						
	(a)	200 km/h	(b)	150 km/h	(c)	$100\sqrt{2}$ km/h	(d)	0			
2.				at an angle θ inimum to maximu			in velocity	. During its path	O		
	(a)	$1:\cos^2\theta$	(b)	$\cos^2 \theta : 1$	(c)	$\cos^3 \theta : 1$	(d)	$1:\cos^3\theta$			
3.		bject is moving all instant. The body	_	ed path for which	its accelerat	tion is $(2\hat{i} + \hat{j})$ m/	s ² and its v	elocity is $(\hat{i} - 3\hat{j})$ 1	m/s		
	(a)	speeding up			(b)) speeding down					
	(c)	moving with u	niform vel	ocity	(d)	moving with un	iform accel	eration			
Max. 1. 2. 3.	Two cars going round curve with speeds one at 90 km/h and other at 15 km/h. Each car experiences same acceleration. The radii of curves are in the ratio of										
	(a)	4:1	(b)	2:1	(c)	16:1	(d)	36:1			
5.						where $\vec{\omega}$ is the ang		ity and \vec{r} is the rad	ius		
	vecto		elocity of a				r = 4j - 3k				
	(a)	$\sqrt{29}$ units	(b)	$\sqrt{31}$ units	(c)	$\sqrt{37}$ units	(d)	$\sqrt{41}$ units			

Space for Rough Work



- 6. A body is moving in a circle with a speed of 1 m/s. This speed increases at a constant rate of 2 m/s every second.

 Assume that the radius of the circle described is 25 m. The total acceleration of the body after 2 s is
 - (a) 2 ms^{-2}
- (b) 25 ms^{-2}
- (c) $\sqrt{5} \text{ ms}^{-2}$
- (d) $\sqrt{7}$ ms⁻²
- 7. A particle is moving on a circular path of radius 'r' with uniform speed 'v'. The change in velocity when the particle moves from P to Q is ($\angle POQ = 40^{\circ}$)

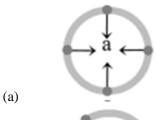


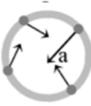
- (a) $2v \cos 40^{\circ}$
- (b) $2v \sin 40^{\circ}$
- (c) $2v \sin 20^{\circ}$
- (d) $2v \cos 20^{\circ}$
- 8. A particle moves in a circle of radius 25 cm at 2 revolutions/s. The acceleration of the particle in m/s² is
 - (a) π^2
- (b) $8\pi^2$
- (c) $4\pi^2$
- (d) $2\pi^2$
- 9. A stone tied to the end of a string 100 cm long is whirled in a horizontal circle with a constant speed. If the stone makes 14 revolutions in 22s, then the acceleration of the stone is x
 - (a) 16 ms^{-2}
- (b) 4 ms^{-2}
- (c) 12 ms^{-2}
- (d) 8 ms^{-2}
- 10. The magnitude of displacement of a particle moving in a circle of radius a with constant angular speed ω varies with time t is
 - (a) 2a sin ωt
- (b) $2a \sin \frac{\omega t}{2}$
- (c) 2a cos ωt
- (d) $2a \cos \frac{\omega t}{2}$

Space for Rough Work



- A car of mass m moves in a horizontal circular path of radius r metre. At an instant its speed is V m/s and is 11. increasing at a rate of a ms⁻². Then the acceleration of the car is
- (b)
- (c) $\sqrt{a^2 + \left(\frac{V^2}{r}\right)^2}$ (d) $\sqrt{a + \frac{V^2}{r}}$
- 12. A car speeds up in a circular path Moving in anticlockwise direction. Which of the following figures illustrates the net acceleration of the car?

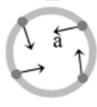




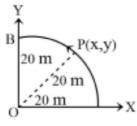
(c)

(b)

(d)



13. A point P moves in a counter-clockwise direction on a circular path as shown in the figure. The movement of P is such that it sweeps out a length $s = t^3 + 5$, where s is in metre and t is in seconds. The radius of the path is 20 m. The magnitude acceleration of P at t = 2 s is:



14 ms⁻² (a)

 $13~\mathrm{ms^{-2}}$ (b)

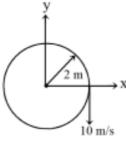
 $12\ ms^{-2}$ (c)

 7.2 ms^{-2} (d)

Space for Rough Work



14. A particle is moving along a circular path in the XY plane. When it crosses the x-axis, it has an acceleration along the path of 1.5 m/s², and is moving with a speed of 10 m/s in the negative y-direction. The total acceleration of the particle is:



- $50\hat{i} 1.5\hat{j} \text{ m/s}^2$ (b) $-50\hat{i} 1.5\hat{j} \text{ m/s}^2$
- (c)
- $10\hat{i} 1.5\hat{j} \,\text{m/s}^2$ (d) $1.5\hat{i} 50\hat{j} \,\text{m/s}^2$
- 15. A particle A moves along a circle of radius of R = 50 cm so that its radius vector r relative to the point O rotates with the constant angular velocity $\omega = 0.40$ rad/s. Then speed of the particle, and the modulus of its total acceleration will be



 $v = 0.4 \text{ m/s}, a = 0.4 \text{ m/s}^2$ (a)

 $v = 0.4 \text{ m/s}, a = 0.32 \text{ m/s}^2$ (b)

 $v = 0.2 \text{ m/s}, a = 0.08 \text{ m/s}^2$ (c)

(d) changing for both v and w w.r.t. time



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JB 2 MR BATCH CHEMISTRY: DCT (SET A)

Topic: Chemical Bonding + Periodic Properties + Mole Concept

16.	According to periodic law of elements, the variation in properties of elements is related to their:											
	(a)	atomic number			(b)	atomic mass						
	(c)	nuclear mass			(d)	neutron-proton ratio						
17.	Which set represents isoelectronic species?											
	(a)	Be, Al ³⁺ , Cl ⁻	(b)	Ca ²⁺ , Cs ⁺ , Br	(c)	Na+, Ca2+, Mg2+	(d)	N^{3-} , F^- , Na^+				
18.	In the periodic table the size of atoms across a period:											
	(a) decreases from right to left					increases from left to right						
	(c)	increases from righ	nt to left		(d)	does not change						
19.	The pair of elements which on combination are most likely to form an ionic compound is:											
	(a)	Na and Ca	(b)	K and O ₂	(c)	O ₂ and Cl ₂	(d)	Al and I ₂				
20.	The correct order of the lattice energies of the following ionic compounds is:											
	(a)	$NaCl > MgBr_2 > CaO > Al_2O_3$				$Al_2O_3\!>MgBr_2\!>\!CaO>NaCl$						
	(c)	$MgBr_2 > Al_2O_3 > CaO > NaCl$				$Al_2O_3\!>CaO>MgBr_2\!>\!NaCl$						
21.	Which	has a giant covalent	structu	re?								
	(a)	PbO ₂	(b)	SiO_2	(c)	NaCl	(d)	AlCl ₃				
22.	Octet rule is not valid for the molecule:											
	(a)	CO_2	(b)	NO	(c)	O_2	(d)	PCl ₃				
23.	Hypervalent compound is:											
	(a)	IF ₇	(b)	NH_3	(c)	BeF_2	(d)	CH ₄				
24.	Which	of the following is r	not isoel	ectronic to others?								
	(a)	NO-	(b)	CN^-	(c)	N_2	(d)	O_2^{2+}				
23.	Octet ru (a) Hyperv (a) Which	ule is not valid for the CO ₂ ralent compound is: IF ₇ of the following is r	he molection (b) (b) not isoel	NO NH ₃ ectronic to others?	(c) (c)	O_2 BeF_2	(d) (d)	PCl ₃				

Space for Rough Work



- 25. Strongest bond is formed by the head on overlapping of:
 - (a) 2s- and 2p- orbitals

(b) 2p- and 2p-orbitals

(c) 2s- and 2s-obritals

- (d) all of these
- 26. Which p-orbital overlapping would give the strongest bond?







(c)

- (d)
- 27. Number of sigma bonds in P₄O₁₀ is:
 - (a) 6
- (b) 7
- (c) 17
- (d) 16

- 28. A sp³-hybrid orbital contains:
 - (a) 1/4 s-character
- (b) 1/2 s-character
- (c) 2/3 s-character
- (d) 3/4 s-character

- 29. Oxidation number of fluorine in F_2O is:
 - (a) +1
- (b) +2
- (c) -1
- (d) -2

 $30. \hspace{1.5cm} Cl_2 + H_2S \rightarrow 2HCl + S,$

In the above reaction, oxidation state of chlorine changes from:

- (a) zero to -1
- (b) 1 to zero
- (c) zero to 1
- (d) remains unchanged

* * * * *





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JB 2 MR BATCH PHYSICS : DCT (SET A) ANSWER KEY

Topics: Circular Motion

1.	(a)	2.	(c)	3.	(b)	4.	(d)	5.	(a)
6.	(c)	7.	(c)	8.	(c)	9.	(a)	10.	(b)
11.	(c)	12.	(d)	13.	(a)	14.	(b)	15.	(b)

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JB 2 MR BATCH CHEMISTRY : DCT (SET A) ANSWER KEY

Topic: Chemical Bonding + Mole Concept + Periodic Properties

16.	(a)	17.	(d)	18.	(c)	19.	(b)	20.	(d)
21.	(b)	22.	(b)	23.	(a)	24.	(a)	25.	(b)
26.	(c)	27.	(d)	28.	(a)	29.	(c)	30.	(a)